

Alien life on Mars - caught in crystals?

Featured scientists: Charles Cockell, UK Centre for Astrobiology, University of Edinburgh, & Nikki Chambers, Astrobiology Teacher, West High School, Torrance, CA

Research Background:

Is there life on other planets besides Earth? This question is not just for science fiction. Scientists are actively exploring the possibility of life beyond Earth. The field of **astrobiology** seeks to understand how life in the universe began and evolved, and whether life exists elsewhere. Our own solar system contains a variety of planets and moons. In recent years scientists have also discovered thousands of planets around stars other than our Sun. So far, none of these places are exactly like Earth. Many planets have environments that would be very difficult for life as we know it to survive. However, there are life forms that exist in extreme environments that we can learn from. On Earth there are extremely hot or acidic environments like volcanic hot springs. Organisms also live in extremely cold places like Antarctic glacier ice. Environments with extremely high pressure, like hydrothermal vents on the ocean floor, also support life. If life can inhabit these extreme environments here on Earth, might extreme life forms exist elsewhere in the universe as well?



A view of the astrobiology lab.

Charles is an astrobiologist from Great Britain who is interested in finding life on other planets. The list of places that we might look for life grows longer every day. Charles thinks that a good place to start is right next door, on our neighboring planet, Mars. We know that Mars currently is cold, dry, and has a very thin atmosphere. Charles is curious to know whether there might still be places on Mars where life could exist, despite its extreme conditions.

While there is no liquid water on the surface of Mars anymore, Mars once had a saltwater ocean covering much of its surface. The conditions on Mars used to be much more like Earth. Liquid water is essential for life as we know it. If there are places on Mars that still hold water, these could be great places to look for evidence of life. Charles thought that perhaps salt crystals, formed when these Martian oceans were evaporating, could trap pockets of liquid water.

Charles and his fellow researcher Nikki knew that there are a number of kinds of salts found in Martian soils, including chlorides, sulfates, perchlorates and others. They

wanted to test their idea that water could get trapped when saltwater with these salts evaporate. They decided to compare the rate of evaporation for solutions with **magnesium sulfate** (MgSO₄) with another common salt solution: **sodium chloride**, or table salt (NaCl). They chose to investigate these two salts because they are less toxic to life as we know it than many of the other chloride, perchlorate, or sulfate salts. Also, from reading the work of other scientists, Charles knows the Martian surface is particularly rich in magnesium sulfate.

Charles and Nikki measured precise quantities of saturated solutions of magnesium sulfate and sodium chloride and placed them into small containers. Plain water was used as a control. There were three replicate containers for each treatment – nine containers in total. They left the containers open to evaporate and recorded their mass daily. They kept collecting data until the mass stopped changing. At this point all of the liquid had evaporated or a salt crust had formed that was impermeable to evaporation. They then compared the final mass of the control containers to the other solutions. They also checked the resulting crusts for the presence or absence of permanent water-containing pockets. Charles and Nikki used these data to determine if either salt makes crystals that can trap water in pockets when it evaporates.

<u>Scientific Question:</u> Do pockets of liquid water form when a salt solution evaporates?

<u>What is the hypothesis?</u> Find the hypothesis in the Research Background and underline it. A hypothesis is a proposed explanation for an observation, which can then be tested with experimentation or other types of studies.

Scientific Data:

Use the images and data table below to answer the scientific question:

LEFT: Sodium chloride crystals, Day 14; little mass change due to evaporation observed. No liquid water is visible.

RIGHT: Magnesium sulfate crystals, Day 14; no further mass change due to evaporation observed. Liquid water is visible in pockets throughout the crystals.





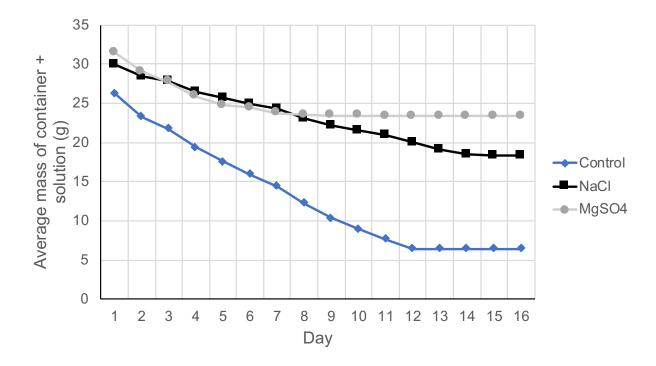
	Control		NaCl	MgSO4	
		Average	Average	Average	
		mass of	mass of	mass of	
		container +	container +	container +	
Date	Day	water (g)	NaCl (g)	MgSO4 (g)	
7/14/2017	1	26.27	29.96	31.45	
7/15/2017	2	23.31	28.50	29.00	
7/16/2017	3	21.74	27.79	27.72	
7/17/2017	4	19.37	26.51	25.90	
7/18/2017	5	17.51	25.62	24.73	
7/19/2017	6	15.90	24.91	24.45	
7/20/2017	7	14.37	24.32	23.75	
7/21/2017	8	12.29	23.10	23.54	
7/22/2017	9	10.32	22.18	23.40	
7/23/2017	10	8.99	21.62	23.42	
7/24/2017	11	7.64	20.99	23.37	
7/25/2017	12	6.45	20.03	23.36	
7/26/2017	13	6.45	19.18	23.36	
7/27/2017	14	6.45	18.45	23.36	
7/28/2017	15	6.45	18.37	23.36	
7/29/2017	16	6.45	18.36	23.36	
*the mass o					

What data will you graph to answer the question?

Independent variable(s):		
Dependent variable(s): _		

Name	!			
Name	<u></u>			

<u>Below is a graph of the data</u>: Identify any changes, trends, or differences you see in your graph. Draw arrows pointing out what you see, and write one sentence describing what you see next to each arrow.



Interpret the data:

Make a claim that answers the scientific question.

What evidence was used to write your claim? Reference specific parts of the table, images, or graph.

Name
ur claim. Connect the data d whether these results
e evidence to explain why or
ess. What new question do be collected to answer your